

Innovative sample preparation technology to reduce bottleneck in a measurement process

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Steps in an analysis

Define the Problem

- What needs to be found? Qualitative and/or quantitative?
- What will the information be used for?
- How accurate and precise does it have to be?
- The analyst (the problem solver) should consult with the client to plan a useful and efficient analysis, including how to obtain a useful sample.

Prepare the Sample for Analysis

- Solid, liquid, or gas?
- Dissolve?
- Ash or digest?
- Distillation

Precipitation

- Solvent extraction
- Solid phase extraction
- Chemical separation or masking of interferences needed?
- Need to concentrate the analyte?
- Need to adjust solution conditions (pH, add reagents)?
- Need to change (derivatize) the analyte for detection?

Perform the Measurement
Calculate the Results and Report

Select a Method

- Sample type
- Size of sample
- Sample preparation needed
- Concentration and range (sensitivity needed)
- Selectivity needed (interferences)
- Accuracy/precision needed
- Tools/instruments available
- Cost
- Speed
- Are methods available in the chemical

literature?

- Are standard methods available?
- Are there regulations that need to be followed?

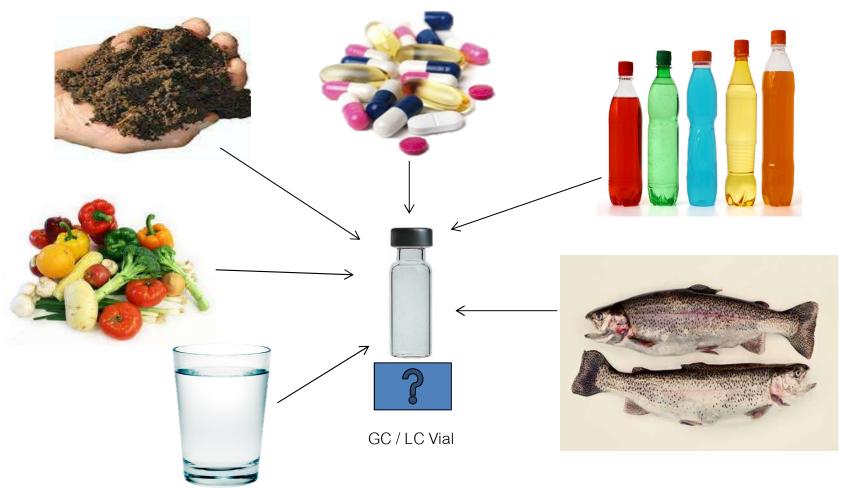
Obtain a Representative Sample

- Sample type/homogeneity/size
- Sampling statistics/errors





The Challenge for Analysis



How do we get analytes out of these samples?





Solvent Extraction

Solvent extraction or Liquid-Liquid Extraction (LLE)



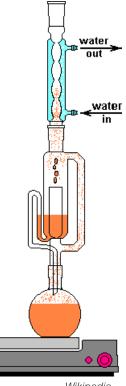
Distribution Coefficient

$$K_D = \frac{[S]_1}{[S]_2}$$

where K_D is the distribution coefficient and the subscripts represent solvent 1 (e.g., an organic solvent) and solvent 2 (e.g., water). If the distribution coefficient is large, the solute will tend to be quantitatively partitioned in solvent 1.

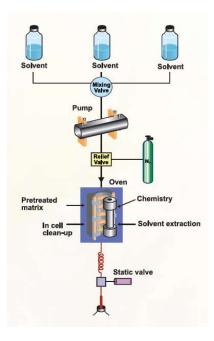
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Soxhlet extraction



Wikipedia

Accelerated and Microwave-Assisted Extraction



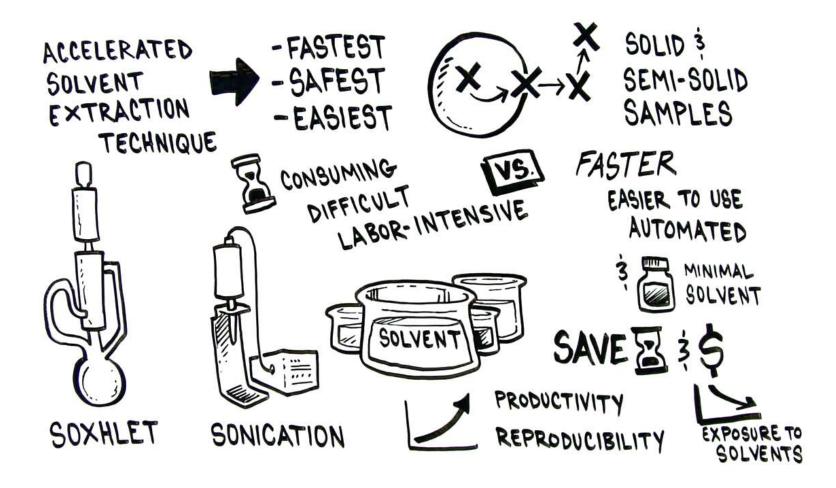
Accelerated solvent extraction is a technique for the efficient extraction of analytes from a solid sample matrix into a solvent. The sample and solvent are placed in a closed vessel and heated to 50 to 200°C. The high pressure allows heating above the boiling point, and the high temperature accelerates the dissolution of analytes in the solvent. Both time of extraction and the volume of solvent needed are greatly reduced over atmospheric extraction.





Accelerated Solvent Extraction

Accelerated Solvent Extraction



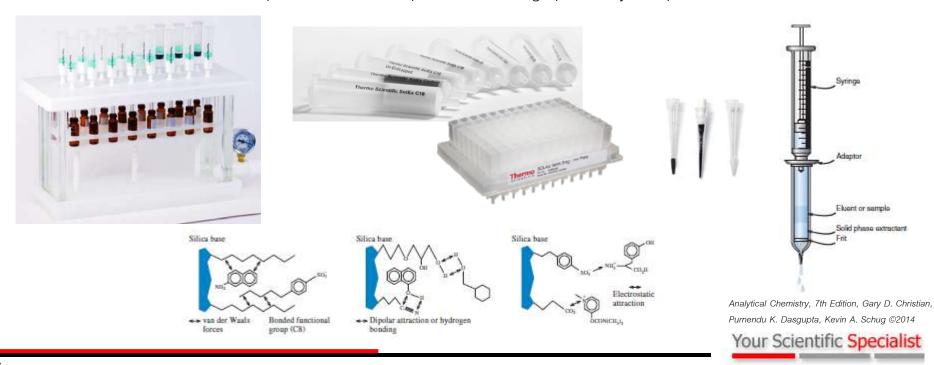


Solid-Phase Extraction

Solid-Phase Extraction

The extracting solvents are limited to those that are water immiscible (for aqueous samples). Emulsions tend to form when the solvents are shaken, and relatively large volumes of solvents are used that generate a substantial waste disposal problem. The operations are often manually performed and may require a back extraction.

Solid-Phase Extraction (SPE), which has become a widely used technique for sample cleanup and concentration prior to chromatographic analysis in particular.

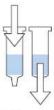




Automated Solid-Phase Extraction

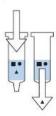
Solid-Phase Extraction

Typical sequence in a solid-phase extraction.



CONDITIONING

Conditioning the sorbent prior to sample application ensures reproducible retention of the compound of interest (the isolate).

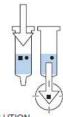


A Rinse the columns to remove other



RETENTION

- Adsorbed isolate
- Undesired matrix constituents
- ▲ Other undesired matrix components



undesired matrix components

- ELUTION ■ Purified and concentrated isolate ready for analysis
- Undesired components remain

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Solid phase extraction is a form of liquid chromatography used in processing samples to selectively isolate constituents of interest from other compounds that may interfere with the analysis.

Before a solid sample can be processed by SPE, it must first be extracted by soxhlet extraction, pressurized fluid extraction (ASE), or other method, to produce a solution that contains the analytes of interest.



More details about SPE

https://www.nist.gov/video/solid-phase-extraction



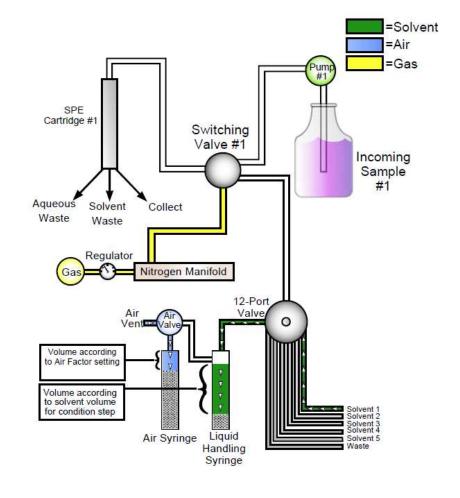


Automated Solid-Phase Extraction

Automated Solid-Phase Extraction











Solvent Evaporation

Solvent Evaporation

After extraction and cleanup process, remaining solvent typically larger than required volume. In order to obtain better sensitivity of analysis, solvent have to be reduced.







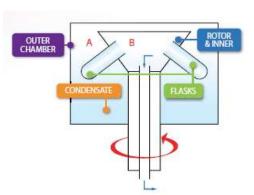


Centrifugal Evaporators

The Rocket Evaporator

Sample evaporation with walkaway capability.











Solid-Phase MicroExtraction

Solid-Phase MicroExtraction (SPME)

SPME is a solvent-free extraction technique, typically used for analyte collection for determination by gas chromatography. The key feature of this device is an extraction fiber, protected inside the needle of a syringe. A typical SPME fiber is made of fused silica coated with a thin layer (7 μ m to 100 μ m thick) of immobilized polymer or a solid adsorbent, or a combination. In a solution or headspace (vapor in equilibrium with the solution in a closed system) analytes are exposed to the fiber and distribute between the sample matrix and the fiber coating during extraction.

Commercial SPME fiber coatings and their applications^a

Fiber coating	Analytes	
olydimethylsiloxane (PDMS) Nonpolar analytes		
Polydimethylsiloxane/ Divinylbenzene (PDMS/ DVB)	Many polar compounds (esp. amines)	
Polyacrylate	Highly polar (ideal for phenols)	
Carboxen/ Polydimethylsiloxane (CAR/PDMS)	Gaseous/volatile analytes	
Carbowax/ Divinylbenzene (CW/DVB)	Polar analytes (esp. alcohols)	
DVB/CAR/PDMS	Broad range of polarities (good for C3-C20 range)	
Carbowax/ Templated resin (CW/TPR)	For HPLC applications	

^aInformation adapted from Supelco application note

"like dissolves like"







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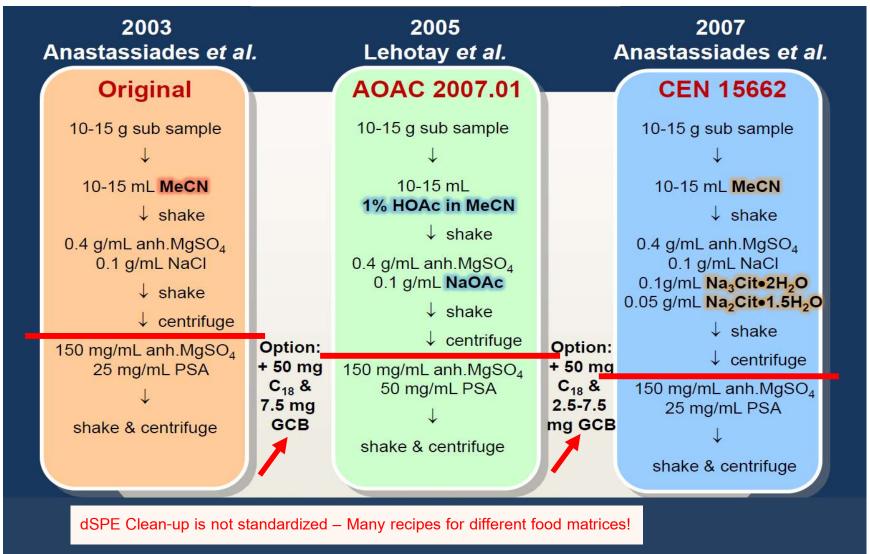
QuEChERS

QuEChERS (pronounced "catchers"), an acronym for Quick, Easy, Cheap, Effective, Rugged and Safe, covers a variety of sample preparation and clean-up techniques for the analysis of multiple pesticide residues in agricultural matrices.





QuEChERS Methods – AOAC vs. CEN



Lehotay, S.J. "Revisiting the Advantages of the QuEChERS Approach to Sample Preparation." Sep Science 2013.

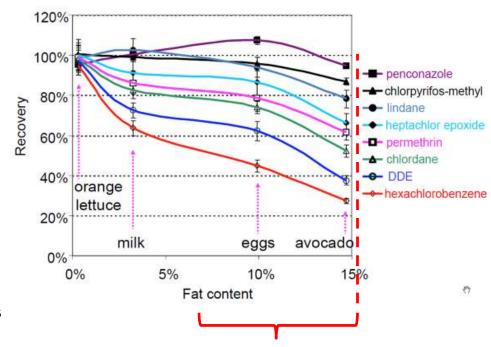


Current Limitations of QuEChERS

As reported by Steven Lehotay

- Too many modified versions
- Cereals require a separate protocol
- Still problems with captan, folpet, captafol
- Spices, tea, and oils give problems
- Matrix effects in complicated matrices
- Low recovery for fatty samples
- Even simpler sample prep possible Solutions by S. Lehotay published:

Use µSPE with PAL3 System autosampler



Recovery drop with increased fat content!





Automated µSPE Clean-up of QuEChERS Extracts

for online GC-MS and LC-MS Multi Residue Pesticide Analysis

Compare to the classical cartridge SPE



Classical SPE

- Limited selectivity
 - High sample and solvent volumes
 - Requires evaporation with N₂
 - End volume >>100 μL in vial
- Vacuum operated
- Drying before elution
- Evaporation step required
 - Sample dilution
- Manual operation
 - Time consuming
 - Low sample throughput
 - Batch processing
- No QA/QC
 - As of manual operation





µSPE

- High selectivity
 - Controlled elution (flow/separation),
 - Sharp elution peak profile, no concentration
 - Final volume $< 100 \mu L$ (or online)
- Positive pressure with liquid syringe
- No drying step
- No evaporation
 - Sample concentration maintained
- Walk away automation
 - Fast with < 10 min
 - High productivity
 - Prep on chromatographic timescale
 - Online to LC-MS and GC-MS
- Traceable
 - Processing well documented





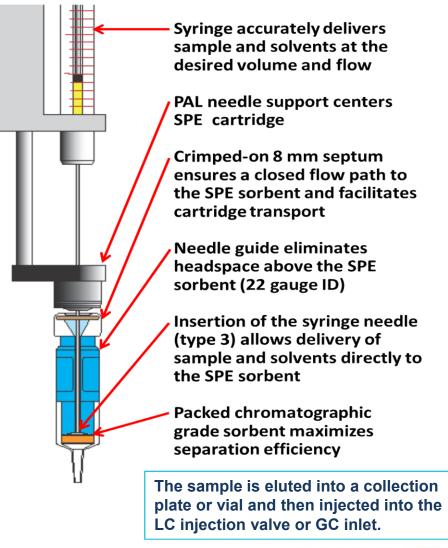


Automated µSPE

Controlled elution from low particle size sorbent bed

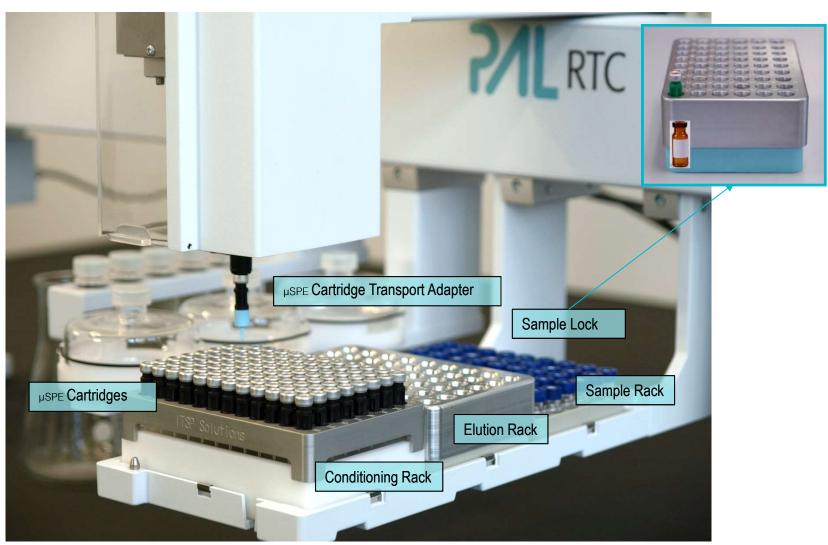
Solvent and sample flow rates are precisely controlled with the syringe plunger.







PAL-RTC µSPE QuEChERS clean-up workflow setup





PAL-RTC µSPE QuEChERS clean-up workflow

A fully automated clean-up workflow

10 g of homog. sample

Add 10 mL Acetonitrile Add ISTD (Triphenylphosphate)

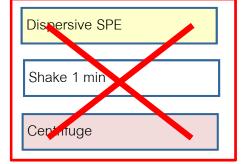
Salt out

Add chemical kit, EN15662

Shake 15 min

Centrifuge

Take 1 mL for µSPE clean-up



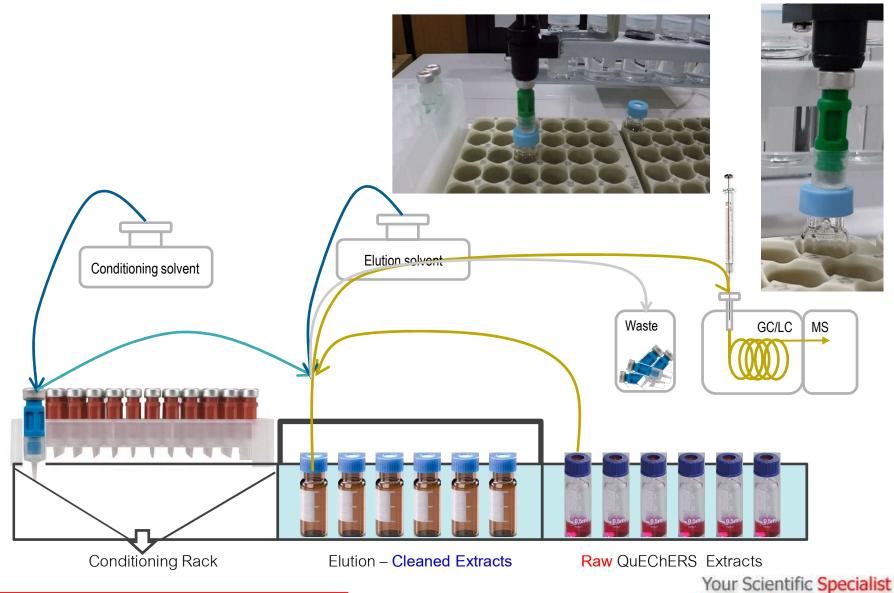
Automated Injection GC-MS and LC-MS





Automated Clean-up Workflow

Step-by-step operation





Automated Clean-up Workflow

PAL3 automated clean-up procedure*

- All steps are program controlled, customizable
- 7 min to injection only Prep-ahead clean-up while previous sample gets analyzed

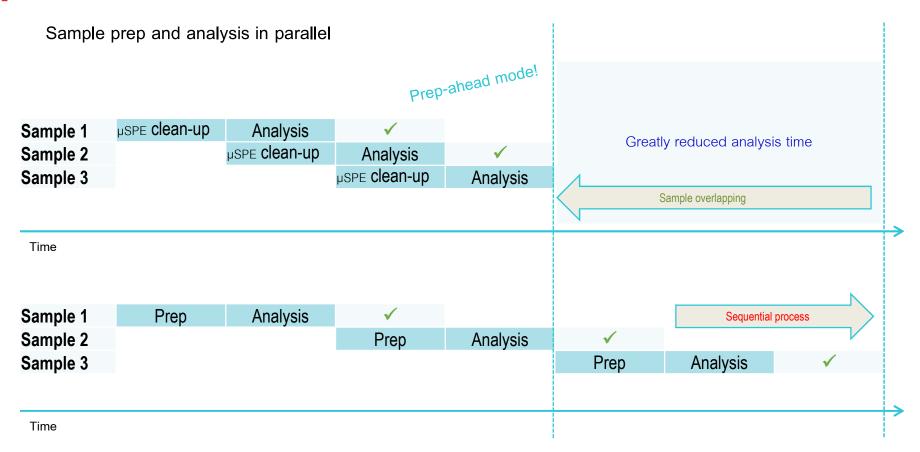
Procedure step*	LCMS	GCMS
Clean syringe with elution solvent		
Condition µSPE cartridge in the conditioning rack	150 µL	200 µL
Transfer cartridge to the elution rack		
Load QuEChERS extract from the sample vial onto the cartridge**	150 µL	100 µL
Clean syringe with elution solvent		
Elute the cartridge with elution solvent	150 µL	150 µL
Collected eluents in 2 mL vial, total volume:	300 μL	250 µL
Discard cartridge to waste baker		
LCMS: Dilute combined extract and mix with syringe	1200 µL	
GCMS: Add analyte protectant solution		30 μL
GCMS: Add Ethyl Acetate and mix with syringe		250 μL
Inject to GCMS or LCMS	10 μL	3 μL

^{*} As of Bruce D. Morris and Richard B. Schriner, J. Agric. Food Chem. 2015, 63, 5107-5119. ** Lehotay Han apply 300 µL raw extract, no separate elution required Your Scientific Specialist



Automated Clean-up Workflow

Prep-ahead for Increased Throughput



The PAL3 performs sample prep and analysis in parallel. As a result, no time is lost in the continuous analysis of samples requiring HS sampling or other time-consuming pretreatments. The MS units works in maximum duty cycle.



Automated QuEChERS Extract Clean-up

References

- Morris, Schriner 2014 Eliminating the need for Matrix-matched
 Calibration Standards_QuEChERS Cleanup Poster
- Morris, Schriner 2015 Automated Column SPE Cleanup of QuEChERS Extracts Using a Zirconia-Based Sorbent for Pesticide Residue Analyses by LC-MS/MS, J Agriculture Food Chem
- Hayward 2016 ITSP Automated Chromatographic SPE using the PAL Autosampler, American Laboratory
- Huebschmann, Boehm Poster EPRW 2016, Limassol, Cyprus
- Huebschmann, Boehm Poster ISCC 2016, Riva del Garda, Italy
- Lehotay et al. 2016 Automated Mini-Column SPE Cleanup for High-Throughput Analysis of Chemical Contaminants in Foods by GC-MS/MS, Chromatographia, DOI 10.1007/s10337-016-3116-y.





Automated workflow for the determination of fatty acid methyl esters (FAME) in fat and fat containing food samples

The analysis of oils, fat and fat containing food via fatty acid methyl esters (FAME) is a common task in governmental, quality control (QC) or contract research laboratories (CRO). Most often the samples are processed manually, which is labor intensive and exposes the lab personnel to potentially hazardous chemicals

การเตรียมตัวอย่างแบบอัตโนมัติ
ด้วยเครื่อง PAL-RTC
สำหรับการวิเคราะห์ FAMEs ใน
ตัวอย่างใขมัน และใขมันในอาหาร
โดยใช้ Sodium-Methoxide
ในกระบวนการ
Transesterification
ก่อนทำการวิเคราะห์ด้วยเครื่อง GC-FID





Automated workflow for the determination of fatty acid methyl esters (FAME) in fat and fat containing food samples



Heat 90 - 100 °C

200 μ L Extracted sample or 20 μ L oil sample +

Methanol: Hexane (4:1) 2 mL + 200 μ L Acetyl Chloride





Cool down at room temperature

Add 6% K₂CO₃ 5 mL

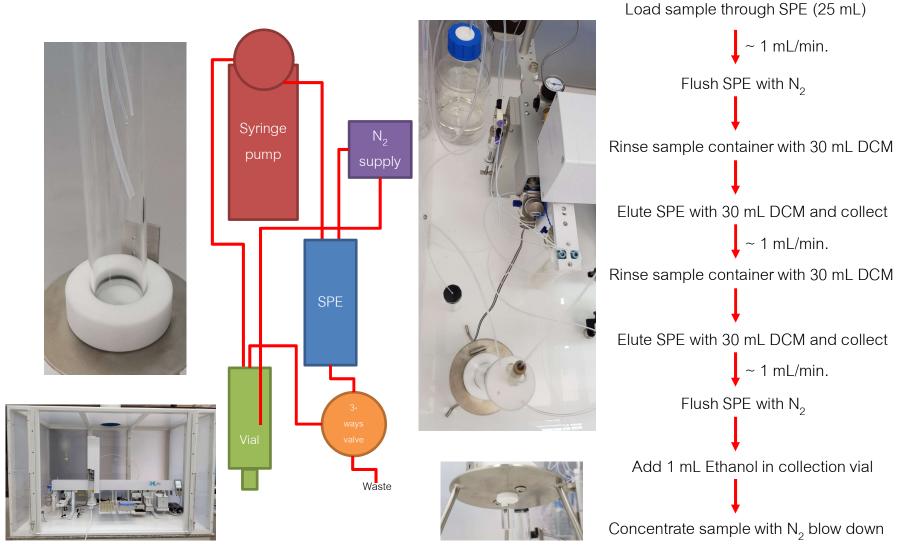
Transfer the supernatant (hexane) to vial

Inject into GC

Your Scientific Specialist



Automated workflow: large volume sample clean up and preconcentration for Nitrosamine determination in rubber samples





Introduction to PAL-RTC for automated sample preparation





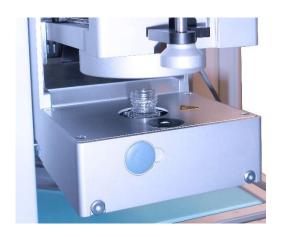
Introduction to PAL-RTC for automated sample preparation

New tools











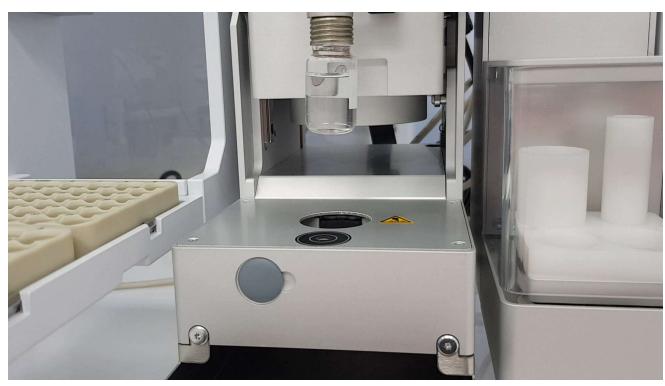






Introduction to PAL-RTC for automated sample preparation

Decapper and Pipette tool in action















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