

Latest Advances in the Analysis of Volatile Organic Compounds by Single Quadrupole GC-MS

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Key Words

- ISQ Single Quadrupole GC-MS
- Environmental Analysis
- EPA Method 524.3
- Single Quadrupole GC-MS
- Volatile Organic Compounds

Introduction

A method was developed for the analysis of volatile organic compounds (VOCs) with the Thermo Scientific ISQ single quadrupole GC-MS using quality control requirements based on United States Environmental Protection Agency (EPA) methodologies. This requires integration of a range of instrumentation, from the sample introduction system to the gas chromatograph and mass spectrometer, to the software for data interpretation, analysis and reporting.¹ While the overall process of analyzing VOC is a mature technique, there are continuous innovations that allow laboratories to meet lower detection limits and analyze new compounds to comply with changing regulations, with higher throughput and improved quality.

Experimental Conditions

For this experiment, a standard GC/MS method for VOC analyses was developed according to published quality control and method guidelines. After establishing a baseline of performance according to these guidelines, improvements to the method were tested by combining changes to the chromatography, taking advantage of the performance capabilities of the hardware, and applying a software package developed around routine environmental GC/MS workflows. By combining these techniques, laboratories can achieve an average decrease of 50% of the analyst review time combined with an increased number of samples that can be analyzed during a 12-hour time period.

The ISQ™ GC-MS (Figure 1) was evaluated at a scanning speed of 2,650 u/sec, which represents 10 full scan mass spectra taken per second (scanning m/z 35 to 300 in 0.1 sec). An OI Eclipse™ 4660 Purge-and-Trap Sample Concentrator equipped with a sample heater and 4551A autosampler were used to deliver 5 mL of sample for analysis. The internal standard and surrogates were added by the Standard Addition Module (SAM) unit. The calibration curve range was 0.4 µg/L to 40 µg/L in the sample. The on column amount for each compound was .05 ng to 5 ng due to the 40 mL/min split flow during the injection.



| Purge and Trap Parameters | Conditions |
|-------------------------------|---|
| Sample Volume: | 5 mL |
| Sample Purge Temperature: | 40 °C |
| Trap: | #10 (Tenax, silica gel, cms) |
| Purge: | 40 mL/min for 11 min |
| Water Management Temperature: | Purge: 110 °C, Desorb: 0 °C, Bake: 240 °C |
| Desorb Preheat Temperature: | 180 °C |
| Desorb Temperature: | 190 °C for 0.5 min |
| Bake Rinse Cycles: | twice |
| Bake Cycle: | 210 °C for 10 min |

| GC-MS Parameters | Conditions |
|-------------------------|--|
| Column: | Thermo Scientific TRACE TR 524 20 m × 0.18 mm, 1.0 µm |
| Inlet Liner: | P and T adapter kit |
| Inlet Temperature: | 175 °C |
| Split Flow: | 40 mL/min |
| Column Flow: | 25 psi at constant pressure |
| GC Temperature Program: | 40 °C for 4 min, 18 °C/min to 100 °C, 40 °C/min to 230 °C for 5 min |
| Solvent Delay: | 0.5 min before activating filament |
| MS Source Temperature: | 250 °C |
| Full Scan: | m/z 35 to 300 |
| Scan Speed: | 2,650 u/sec (0.1 sec) |

Table 1: Instrument parameters



Figure 1: ISQ GC-MS

Results and Discussion

Peak Shape

EPA Method 524.3 recommended, and allowable, range of values for the purge and trap parameters reduces the cycle time of the purge and trap and minimize the amount of water injected. The shorter desorb time and rapid oven temperature program on a narrow-bore 0.18 mm ID capillary column gives an analysis time of 15 minutes (Figure 2). Improvement in the carrier gas control reduced

band broadening of the first 6 gases, resulting in more Gaussian peak shapes (Figure 3). A scan rate of 2,650 u/sec offered excellent precision and was compatible for use with the faster chromatographic method. This, combined with shortened data review times, leads to an increase in productivity as shown in Figure 4.

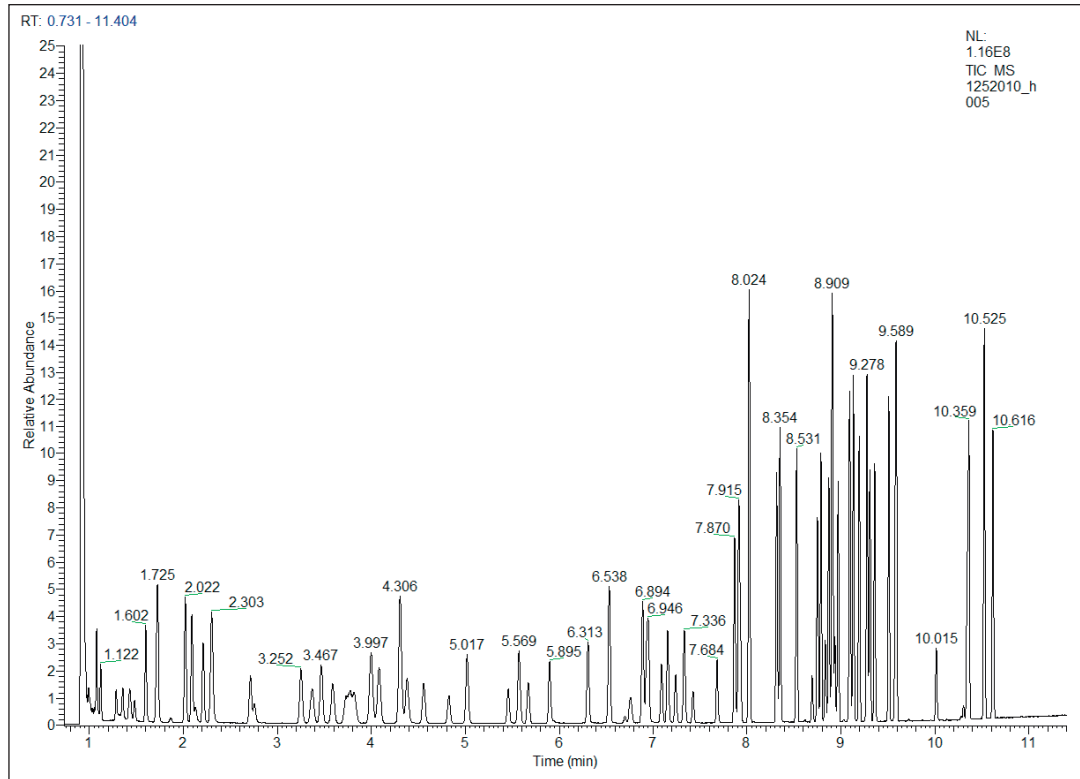


Figure 2:
TIC of 20 µg/L
standard

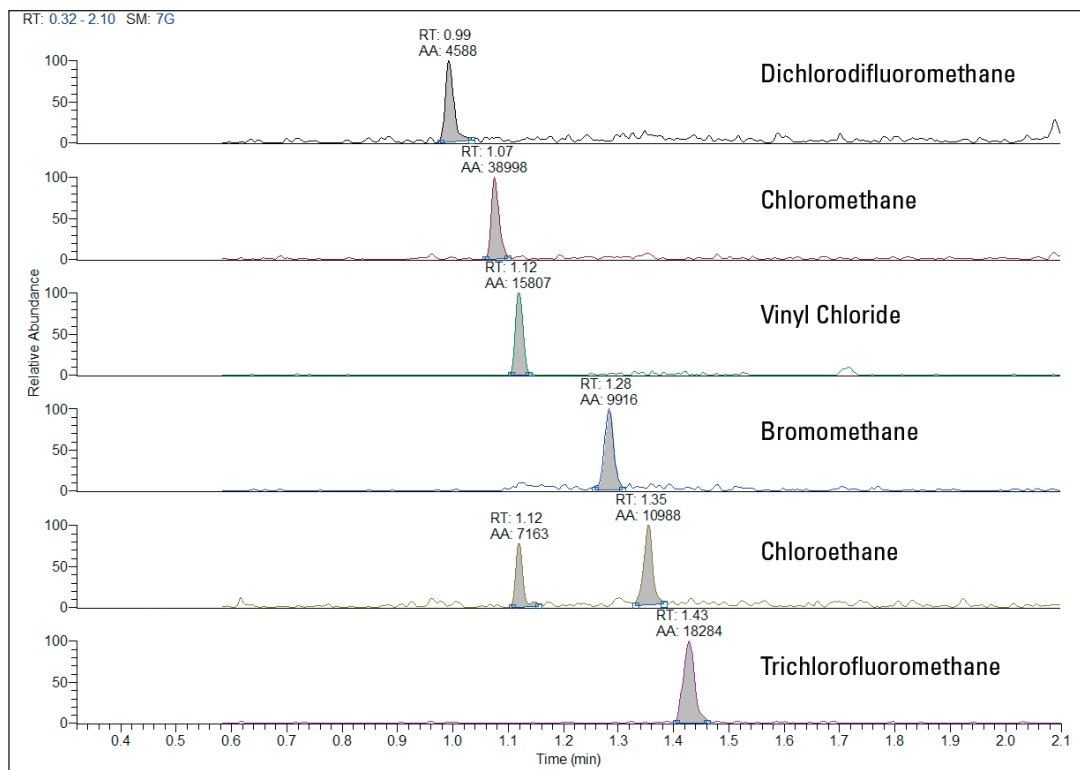


Figure 3:
Extracted ion
profile of
first 6 gases

Calibration Curve

A calibration curve was made from 0.4 to 40 ppb. A 5 mL sample was purged onto a trap using the OI Eclipse 4660 and 4551 liquid autosampler. The internal standard (fluorobenzene) and surrogates (4-bromofluorobenzene and 1,2-dichlorobenzene-d4) were added automatically by the SAM at a final concentration of 5 µg/L. The average %RSD of all compounds was 8 %RSD and the internal standard fluorobenzene was 6 %RSD. The results are shown in Table 2.

Method Detection Limits

Method Detection Limits (MDLs) were generated by analyzing seven replicates at 0.4 µg/L in organic free water or .05 ng to the mass spectrometer. The average MDL was 0.104 µg/L. The results are shown in Table 3.

Conclusions

Improvements to the method were tested by combining changes to the chromatography, employing the capabilities of the hardware, and applying a software package developed around routine GC/MS workflows. By combining these techniques, laboratories can achieve an average decrease of 50% of the analyst review time combined with an increased number of samples that can

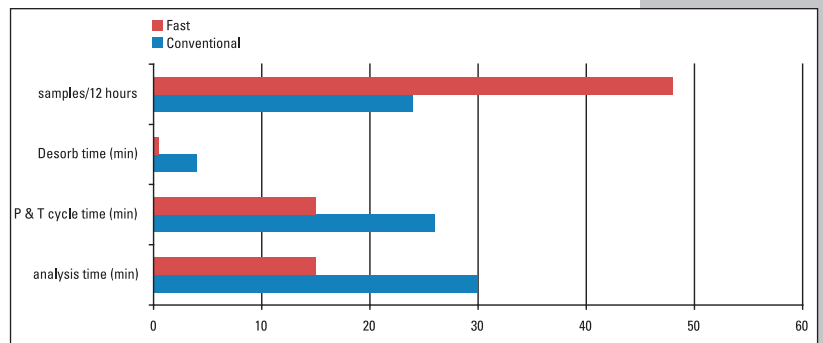


Figure 4: Reduction of analysis time

be analyzed during a 12-hour time period while also attaining lower MDLs. All of these improvements resulted in doubling the throughput of samples analyzed in a 12-hour time period – from 24 to 48 samples. The ISQ single quadrupole GC-MS was evaluated and found to provide good sensitivity, spectral purity, and linear dynamic range for the method.

References

1. EPA Method 524.3 Measurement of Purgeable Organic Compounds in Water by Capillary Column gas Chromatography/Mass Spectrometry, U.S. Environmental Protection Agency, Cincinnati, OH, Version 1, June 2009

| % RSD | | % RSD | | % RSD | |
|------------------------------------|----|-----------------------------------|----|-------------------------------------|----|
| dichlorodifluoromethane | 10 | methacrylonitrile | 9 | styrene | 8 |
| chloromethane | 8 | 1,2-dichloroethane | 7 | isopropylbenzene | 10 |
| vinylchloride | 8 | fluorobenzene (ISTD) | 6 | 4-bromofluorobenzene (surr) | 5 |
| bromomethane | 10 | trichloroethylene | 5 | bromobenzene | 4 |
| chloroethane | 9 | dibromomethane | 10 | n-propylbenzene | 5 |
| trichlorofluoromethane | 10 | 1,2-dichloropropane | 6 | 1,1,2,2-tetrachloroethane | 7 |
| diethylether | 7 | bromodichloromethane | 6 | 2-chlorotoluene | 3 |
| 1,1-dichloroethylene | 7 | methyl methacrylate | 9 | 1,2,3-trichloropropane | 4 |
| carbon disulfide | 14 | <i>cis</i> -1,3-dichloropropene | 9 | 1,3,5-trimethylbenzene | 7 |
| allyl chloride | 6 | toluene | 7 | <i>trans</i> -1,4-dichloro-2-butene | 5 |
| methylene chloride | 7 | chloroacetonitrile | 5 | 4-chlorotoluene | 8 |
| <i>trans</i> -1,2-dichloroethylene | 10 | 1,1-dichloro-2-propanone | 8 | <i>p</i> -isopropyltoluene | 8 |
| methyl tert-butyl ether | 6 | tetrachloroethylene | 10 | tert-butylbenzene | 11 |
| 1,1-dichloroethane | 5 | 2-nitropropane | 9 | 1,2,4-trimethylbenzene | 7 |
| acrylonitrile | 11 | 4-methyl-2-pentanone | 9 | pentachloroethane | 4 |
| <i>cis</i> -1,2-dichloroethylene | 7 | <i>trans</i> -1,3-dichloropropene | 10 | sec-butylbenzene | 12 |
| 2,2-dichloropropane | 13 | 1,1,2-trichloroethane | 9 | 1,3-dichlorobenzene | 12 |
| bromochloromethane | 5 | ethyl methacrylate | 6 | 1,4-dichlorobenzene | 8 |
| chloroform | 6 | dibromochloromethane | 11 | n-butylbenzene | 3 |
| carbon tetrachloride | 10 | 1,3-dichloropropane | 9 | hexachloroethane | 10 |
| tetrahydrofuran | 7 | 1,2-dibromoethane | 8 | 1,2-dichlorobenzene | 9 |
| methyl acrylate | 9 | 2-hexanone | 5 | 1,2-dichlorobenzene d4 (surr) | 6 |
| 1,1,1-trichloroethane | 5 | chlorobenzene | 8 | 1,2-dibromo-3-chloropropane | 5 |
| 1,1-dichloropropylene | 7 | ethylbenzene | 6 | nitrobenzene | 10 |
| 1-chlorobutane | 9 | 1,1,1,2-tetrachloroethane | 5 | hexachlorobutadiene | 8 |
| 2-butanone | 9 | m&p-xylene | 5 | 1,2,4-trichlorobenzene | 12 |
| benzene | 6 | o-xylene | 6 | naphthalene | 10 |
| propionitrile | 11 | bromoform | 4 | 1,2,3-trichlorobenzene | 6 |

Table 2: Calibration curve results (0.4, 2, 4, 20 µg/L)

| | Avg Conc (µg/L) | Std Dev | % RSD | MDL (µg/L) | | Avg Conc (µg/L) | Std Dev | % RSD | MDL (µg/L) |
|----------------------------|-----------------|---------|-------|------------|-------------------------------|-----------------|---------|-------|------------|
| dichlorodifluoromethane | 0.34 | 0.016 | 5 | 0.051 | 2-nitropropane | 0.41 | 0.035 | 8 | 0.109 |
| chloromethane | 0.43 | 0.028 | 7 | 0.087 | 4-methyl-2-pentanone | 0.41 | 0.031 | 8 | 0.098 |
| vinylchloride | 0.4 | 0.02 | 5 | 0.062 | trans-1,3-dichloropropene | 0.38 | 0.021 | 5 | 0.065 |
| bromomethane | 0.45 | 0.047 | 11 | 0.148 | 1,1,2-trichloroethane | 0.4 | 0.022 | 5 | 0.069 |
| chloroethane | 0.45 | 0.077 | 17 | 0.242 | ethyl methacrylate | 0.37 | 0.022 | 6 | 0.069 |
| trichlorofluoromethane | 0.41 | 0.039 | 9 | 0.121 | dibromochloromethane | 0.37 | 0.027 | 7 | 0.084 |
| diethylether | 0.45 | 0.031 | 7 | 0.097 | 1,3-dichloropropane | 0.38 | 0.019 | 5 | 0.059 |
| 1,1-dichloroethylene | 0.42 | 0.049 | 12 | 0.154 | 1,2-dibromoethane | 0.38 | 0.019 | 5 | 0.061 |
| carbon disulfide | 0.43 | 0.032 | 7 | 0.099 | 2-hexanone | 0.47 | 0.061 | 13 | 0.193 |
| allyl chloride | 0.42 | 0.041 | 10 | 0.13 | chlorobenzene | 0.39 | 0.016 | 4 | 0.052 |
| methylene chloride | 0.46 | 0.032 | 7 | 0.101 | ethylbenzene | 0.4 | 0.019 | 5 | 0.059 |
| acetone* | 4.79 | 0.399 | 8 | 1.255 | 1,1,1,2-tetrachloroethane | 0.42 | 0.038 | 9 | 0.119 |
| trans-1,2-dichloroethylene | 0.41 | 0.046 | 11 | 0.144 | m&p-xylene | 0.79 | 0.041 | 5 | 0.13 |
| methyl tert-butyl ether | 0.42 | 0.019 | 4 | 0.058 | o-xylene | 0.4 | 0.009 | 2 | 0.03 |
| 1,1-dichloroethane | 0.42 | 0.027 | 7 | 0.085 | bromoform | 0.37 | 0.047 | 13 | 0.147 |
| acrylonitrile | 0.41 | 0.036 | 9 | 0.115 | styrene | 0.39 | 0.017 | 5 | 0.055 |
| cis-1,2-dichloroethylene | 0.41 | 0.029 | 7 | 0.092 | isopropylbenzene | 0.39 | 0.018 | 5 | 0.057 |
| 2,2-dichloropropane | 0.38 | 0.026 | 7 | 0.083 | 4-bromofluorobenzene (surr) | 5.27 | 0.234 | 4 | 0.735 |
| bromochloromethane | 0.39 | 0.042 | 11 | 0.131 | bromobenzene | 0.4 | 0.029 | 7 | 0.092 |
| chloroform | 0.41 | 0.021 | 5 | 0.066 | n-propylbenzene | 0.39 | 0.02 | 5 | 0.064 |
| carbon tetrachloride | 0.42 | 0.03 | 7 | 0.096 | 1,1,2,2-tetrachloroethane | 0.38 | 0.022 | 6 | 0.069 |
| tetrahydrofuran | 0.48 | 0.087 | 18 | 0.272 | 2-chlorotoluene | 0.4 | 0.022 | 6 | 0.069 |
| methyl acrylate | 0.39 | 0.041 | 11 | 0.128 | 1,2,3-trichloropropane | 0.41 | 0.017 | 4 | 0.053 |
| 1,1,1-trichloroethane | 0.43 | 0.035 | 8 | 0.112 | 1,3,5-trimethylbenzene | 0.39 | 0.018 | 5 | 0.057 |
| 1,1-dichloropropylene | 0.39 | 0.03 | 8 | 0.093 | trans-1,4-dichloro-2-butene | 0.4 | 0.025 | 6 | 0.078 |
| 2-butanone | 0.52 | 0.099 | 19 | 0.31 | 4-chlorotoluene | 0.36 | 0.014 | 4 | 0.045 |
| 1-chlorobutane | 0.35 | 0.096 | 28 | 0.302 | p-isopropyltoluene | 0.36 | 0.014 | 4 | 0.045 |
| benzene | 0.4 | 0.011 | 3 | 0.036 | tert-butylbenzene | 0.36 | 0.012 | 3 | 0.039 |
| propionitrile | 0.43 | 0.062 | 14 | 0.194 | 1,2,4-trimethylbenzene | 0.4 | 0.017 | 4 | 0.053 |
| methacrylonitrile | 0.42 | 0.053 | 13 | 0.166 | pentachloroethane | 0.4 | 0.051 | 13 | 0.16 |
| 1,2-dichloroethane | 0.4 | 0.024 | 6 | 0.076 | sec-butylbenzene | 0.39 | 0.017 | 4 | 0.053 |
| fluorobenzene (ISTD) | | | 5 | | 1,3-dichlorobenzene | 0.44 | 0.013 | 3 | 0.042 |
| trichloroethylene | 0.41 | 0.025 | 6 | 0.077 | 1,4-dichlorobenzene | 0.52 | 0.019 | 4 | 0.061 |
| dibromomethane | 0.39 | 0.05 | 13 | 0.156 | n-butylbenzene | 0.39 | 0.017 | 4 | 0.054 |
| 1,2-dichloropropane | 0.4 | 0.026 | 7 | 0.082 | hexachloroethane | 0.37 | 0.019 | 5 | 0.059 |
| bromodichloromethane | 0.38 | 0.024 | 6 | 0.075 | 1,2-dichlorobenzene | 0.42 | 0.025 | 6 | 0.078 |
| methyl methacrylate | 0.38 | 0.032 | 8 | 0.099 | 1,2-dichlorobenzene d4 (surr) | 5.44 | 0.435 | 8 | 1.369 |
| cis-1,3-dichloropropene | 0.38 | 0.024 | 6 | 0.077 | 1,2-dibromo-3-chloropropane | 0.38 | 0.046 | 12 | 0.144 |
| toluene | 0.38 | 0.008 | 2 | 0.025 | nitrobenzene | 0.02 | 0.046 | 19 | 0.144 |
| chloroacetonitrile | 0.31 | 0.086 | 28 | 0.27 | hexachlorobutadiene | 0.39 | 0.039 | 10 | 0.122 |
| 1,1-dichloro-2-propanone | 0.35 | 0.082 | 23 | 0.257 | 1,2,4-trichlorobenzene | 0.42 | 0.023 | 5 | 0.071 |
| tetrachloroethylene | 0.39 | 0.079 | 20 | 0.248 | naphthalene | 0.38 | 0.012 | 3 | 0.038 |
| * lab air contamination | | | | | 1,2,3-trichlorobenzene | 0.42 | 0.025 | 6 | 0.079 |

Table 3: Method detection limits (7 replicates at 0.4 µg/L)

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