

# Innovative sample preparation technology to reduce bottleneck in a measurement process

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### Steps in an analysis

#### Define the Problem

- What needs to be found? Qualitative and/or quantitative?
- What will the information be used for?
- How accurate and precise does it have to be?
- The analyst (the problem solver) should consult with the client to plan a useful and efficient analysis, including how to obtain a useful sample.

### Prepare the Sample for Analysis

- Solid, liquid, or gas?
- Dissolve?

• Ash or digest?

Distillation

Precipitation

- Solvent extraction
- Solid phase extraction
- Chemical separation or masking of interferences needed?
- Need to concentrate the analyte?
- Need to adjust solution conditions (pH, add reagents)?
- Need to change (derivatize) the analyte for detection?

#### Perform the Measurement

Calculate the Results and Report

### Select a Method

- Sample type
- Size of sample
- Sample preparation needed
- Concentration and range (sensitivity needed)
- Selectivity needed (interferences)
- Accuracy/precision needed
- Tools/instruments available
- Cost
- Speed
- Are methods available in the chemical literature?
- Are standard methods available?
- Are there regulations that need to be followed?

### Obtain a Representative Sample

- Sample type/homogeneity/size
- Sampling statistics/errors



### The Challenge for Analysis



How do we get analytes out of these samples?





### Solvent Extraction

Solvent extraction or Liquid-Liquid Extraction (LLE)

#### Accelerated and Microwave-Assisted Extraction



### **Distribution Coefficient**



where K<sub>D</sub> is the distribution coefficient and the subscripts represent solvent 1 (e.g., an organic solvent) and solvent 2 (e.g., water). If the distribution coefficient is large, the solute will tend to be quantitatively partitioned in solvent 1. *Analytical Chemistry, 7th Edition, Gary D. Christian, Purnendu K. Dasgupta, Kevin A. Schug ©2014* 

### Soxhlet extraction





Accelerated solvent extraction is a technique for the efficient extraction of analytes from a solid sample matrix into a solvent. The sample and solvent are placed in a closed vessel and heated to 50 to 200°C. The high pressure allows heating above the boiling point, and the high temperature accelerates the dissolution of analytes in the solvent. Both time of extraction and the volume of solvent needed are greatly reduced over atmospheric extraction.





### **Accelerated Solvent Extraction**

Accelerated Solvent Extraction





#### Solid-Phase Extraction

The extracting solvents are limited to those that are water immiscible (for aqueous samples). Emulsions tend to form when the solvents are shaken, and relatively large volumes of solvents are used that generate a substantial waste disposal problem. The operations are often manually performed and may require a back extraction.

Solid-Phase Extraction (SPE), which has become a widely used technique for sample cleanup and concentration prior to chromatographic analysis in particular.





# **Automated Solid-Phase Extraction**

#### Solid-Phase Extraction



Solid phase extraction is a form of liquid chromatography used in processing samples to selectively isolate constituents of interest from other compounds that may interfere with the analysis. Before a solid sample can be processed by SPE, it must first be extracted by soxhlet extraction, pressurized fluid extraction (ASE), or other method, to produce a solution that contains the analytes of interest.



More details about SPE

https://www.nist.gov/video/solid-phase-extraction





### **Automated Solid-Phase Extraction**

Automated Solid-Phase Extraction









### Solvent Evaporation

#### Solvent Evaporation

After extraction and cleanup process, remaining solvent typically larger than required volume. In order to obtain better sensitivity of analysis, solvent have to be reduced.











# **Centrifugal Evaporators**

The Rocket Evaporator













#### Solid-Phase MicroExtraction (SPME)

SPME is a solvent-free extraction technique, typically used for analyte collection for determination by gas chromatography. The key feature of this device is an extraction fiber, protected inside the needle of a syringe. A typical SPME fiber is made of fused silica coated with a thin layer (7  $\mu$ m to 100  $\mu$ m thick) of immobilized polymer or a solid adsorbent, or a combination. In a solution or headspace (vapor in equilibrium with the solution in a closed system) analytes are exposed to the fiber and distribute between the sample matrix and the fiber coating during extraction.

Commercial SPME fiber coatings and their applications <sup>a</sup>		
Fiber coating	Analytes	
Polydimethylsiloxane (PDMS)	Nonpolar analytes	
Polydimethylsiloxane/ Divinylbenzene (PDMS/ DVB)	Many polar compounds (esp. amines)	
Polyacrylate	Highly polar (ideal for phenols)	
Carboxen/ Polydimethylsiloxane (CAR/PDMS)	Gaseous/volatile analytes	
Carbowax/ Divinylbenzene (CW/DVB)	Polar analytes (esp. alcohols)	
DVB/CAR/PDMS	Broad range of polarities (good for C3-C20 range)	
Carbowax/ Templated resin (CW/TPR)	For HPLC applications	







aInformation adapted from Supelco application note

Analytical Chemistry, 7th Edition, Gary D. Christian, Purnendu K. Dasgupta, Kevin A. Schug ©2014

"like dissolves like"







### Solid-Phase MicroExtraction

HiSorb, sorptive extraction



An innovative, labour-saving sampling system for the analysis of volatile and semi-volatile organic compounds (VOCs and SVOCs) in liquids and solids by TD–GC–MS.





Simple workflow for maximum productivity



Probe insertion: Two probe lengths allow immersive or headspace sampling in 20 or 10 mL vials.



Analyte extraction: The HiSorb Agitator efficiently mixes and heats the sample.



Probe washing: Probes are washed and dried to remove residual matrix.



Analysis: The HiSorb probe is inserted into a standard TD tube for analysis by TD–GC–MS.





Fast and flexible sampling of chemicals and odours released from materials and foods

Simultaneously collect volatile and semi-volatile organic compounds (VOCs and SVOCs) from up to six samples





# **Thermal Desorption**

Thermal desorption (TD) is the process of heating a material to release adsorbed compounds from it.

As an analytical method, TD is used as a pre-concentration technique for gas chromatography (GC), making GC compatible with low-concentration analytes that would otherwise be impossible to detect with this method.

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Tube desorption and inlet split

FOODSINGTON DESCONTINUE TO CHART

UNITY XT

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Sample tube heated in a flow of carrier gas and analytes swept onto an electrically cooled focusing trap, typically held between Ambient and –30°C.

Focusing trap rapidly heated (up to 100°C/s) in a reverse flow of carrier gas ('backflush' operation), to transfer the analytes to the GC column.

Trap desorption and outlet split



# Thermal Desorption (Direct desorption)

Direct desorption (Dynamic Headspace)

Provides a 'gas extraction' or 'dynamic headspace' alternative to conventional solvent extraction







# QuEChERS

QuEChERS (pronounced "catchers"), an acronym for Quick, Easy, Cheap, Effective, Rugged and Safe, covers a variety of sample preparation and clean-up techniques for the analysis of multiple pesticide residues in agricultural matrices.





# QuEChERS Methods – AOAC vs. CEN





# **Current Limitations of QuEChERS**

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- Too many modified versions
- Cereals require a separate protocol
- Still problems with captan, folpet, captafol
- Spices, tea, and oils give problems
- Matrix effects in complicated matrices
- Low recovery for fatty samples
- Even simpler sample prep possible Solutions by S. Lehotay published:

Use  $\mu\text{SPE}$  with PAL3 System autosampler



Recovery drop with increased fat content!

Lehotay, S.J. "Revisiting the Advantages of the QuEChERS Approach to Sample Preparation." Sep Science 2013.



# Automated µSPE Clean-up of QuEChERS Extracts

for online GC-MS and LC-MS Multi Residue Pesticide Analysis

### Compare to the classical cartridge SPE

#### **Classical SPE**

- Limited selectivity
  - High sample and solvent volumes
  - Requires evaporation with N<sub>2</sub>
  - End volume >>100 μL in vial
- Vacuum operated
- Drying before elution
- Evaporation step required
  - Sample dilution
- Manual operation
  - Time consuming
  - Low sample throughput
  - Batch processing
- No QA/QC
  - As of manual operation

#### μSPE

- High selectivity
  - Controlled elution (flow/separation),
  - Sharp elution peak profile, no concentration
  - Final volume < 100  $\mu$ L (or online)
- Positive pressure with liquid syringe
- No drying step
- No evaporation
  - Sample concentration maintained
- Walk away automation
  - Fast with < 10 min</li>
  - High productivity
  - Prep on chromatographic timescale
  - Online to LC-MS and GC-MS
- Traceable
  - Processing well documented



Ingenious sample handling





# Automated µSPE









### A fully automated clean-up workflow





# Automated Clean-up Workflow

Step-by-step operation





# Automated Clean-up Workflow

### PAL3 automated clean-up procedure\*

- All steps are program controlled, customizable
- 7 min to injection only Prep-ahead clean-up while previous sample gets analyzed

Procedure step*	LCMS	GCMS
Clean syringe with elution solvent		
Condition µSPE cartridge in the conditioning rack	150 µL	200 µL
Transfer cartridge to the elution rack		
Load QuEChERS extract from the sample vial onto the cartridge**	150 µL	100 µL
Clean syringe with elution solvent		
Elute the cartridge with elution solvent	150 µL	150 µL
Collected eluents in 2 mL vial, total volume:	300 µL	250 µL
Discard cartridge to waste baker		
LCMS: Dilute combined extract and mix with syringe	1200 µL	
GCMS: Add analyte protectant solution		30 µL
GCMS: Add Ethyl Acetate and mix with syringe		250 µL
Inject to GCMS or LCMS	10 µL	3 µL

\* As of Bruce D. Morris and Richard B. Schriner, J. Agric. Food Chem. 2015, 63, 5107-5119. \*\* Lehotay Han apply 300 µL raw extract, no separate elution required Your Scientific Specialist



# Automated Clean-up Workflow

Prep-ahead for Increased Throughput



The PAL3 performs sample prep and analysis in parallel. As a result, no time is lost in the continuous analysis of samples requiring HS sampling or other time-consuming pretreatments. The MS units works in maximum duty cycle.



### Automated QuEChERS Extract Clean-up

### References

- Morris, Schriner 2014 Eliminating the need for Matrix-matched Calibration Standards QuEChERS Cleanup Poster
- Morris, Schriner 2015 Automated Column SPE Cleanup of QuEChERS Extracts Using a Zirconia-Based Sorbent for Pesticide Residue Analyses by LC-MS/MS, J Agriculture Food Chem
- Hayward 2016 ITSP Automated Chromatographic SPE using the PAL Autosampler, American Laboratory
- Huebschmann, Boehm Poster EPRW 2016, Limassol, Cyprus
- Huebschmann, Boehm Poster ISCC 2016, Riva del Garda, Italy
- Lehotay et al. 2016 Automated Mini-Column SPE Cleanup for High-Throughput Analysis of Chemical Contaminants in Foods by GC-MS/MS, Chromatographia, DOI 10.1007/s10337-016-3116-y.





The analysis of oils, fat and fat containing food via fatty acid methyl esters (FAME) is a common task in governmental, quality control (QC) or contract research laboratories (CRO). Most often the samples are processed manually, which is labor intensive and exposes the lab personnel to potentially hazardous chemicals

การเตรียมตัวอย่างแบบอัตโนมัติ ด้วยเครื่อง PAL-RTC สำหรับการวิเคราะห์ FAMEs ใน ตัวอย่างไขมัน และไขมันในอาหาร โดยใช้ Sodium-Methoxide ในกระบวนการ Transesterification ก่อนทำการวิเคราะห์ด้วยเครื่อง GC-FID



Automated workflow for the determination of fatty acid

methyl esters (FAME) in fat and fat containing food samples



200  $\mu$ L Extracted sample or 20  $\mu$ L oil sample + Methanol : Hexane (4:1) 2 mL + 200  $\mu$ L Acetyl Chloride





Heat 90 – 100 °C



Automated workflow: large volume sample clean up and

Spec preconcentration for Nitrosamine determination in rubber samples









New tools















Decapper and Pipette tool in action



Liquid

Extraction













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