



## UHPLC-MS Basic Principles and Applications

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Product Specialist LC/MS

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## Fundamental of Liquid Chromatography



Sci Spec

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https://www.thermofisher.com/order/catalog/product/IQLAAAGABHFAPUMZZZ?SID=srch-srp-IQLAAAGABHFAPUMZZZ

#### Sci Spec

## HPLC System Range



## **Sci** Thermo Analytical L6 Systems



## Vanquish<sup>™</sup> Max Pressure 1517 bar

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mo





Higher resolution – narrower peaks Higher sensitivity – taller peaks Higher peak capacity (more peaks / unit time) – narrower peaks

#### Increase Speed, Maintain Resolution 200x2.1mm Jec Speeding up analysis with 1.9 mm Hypersil GOLD 600µl/min 655 bar 1.9**m** 400µl/min 190 bar 3**m**M Speed 250µl/min 102 bar 5**m**M 150µl/min 68 bar 8**m** 100µl/min 56bar 12**m**M 10 12 0 2 18 4 6 8 14 16 Time (min)



#### The UltiMate<sup>™</sup> 3000 LC Systems









#### **Sci** Fundamental of Mass Spectrometry Your Scientific Specialist

https://www.thermofisher.com/order/catalog/product/TSQ02-10001?SID=srch-srp-TSQ02-10001



"The basis in mass spectrometry (MS) is the production of ions, that are subsequently separated or filtered according to their mass-to-charge (m/z) ratio, and detected. The resulting mass spectrum is a plot of the (relative) abundance of the produced ions as a function of the m/z ratio."

Niessen, W. M. A.; Van der Greef, J., *Liquid Chromatography–Mass Spectrometry: Principles and Applications*, 1992, Marcel Dekker, Inc., New York, p. 29.



## Information Rich Data





#### • Pharmaceutical analysis

- Bioavailability studies
- Drug metabolism studies, pharmacokinetics
- Characterization of potential drugs
- Drug degradation product analysis
- Screening of drug candidates
- Identifying drug targets

- Biomolecule
   characterization
  - Proteins and peptides
  - Oligonucleotides
- Environmental analysis
  - Pesticides on foods
  - Soil and groundwater contamination
- Forensic analysis/clinical





• Operate at very low pressure (10<sup>-5</sup> to 10<sup>-7</sup> torr)

(Atmosphere = 760 torr)

- Mass spectrometer work with IONS
- •Measure gas-phase ions
- Determine the mass are separated according to their mass-to-charge (m/z) ratio





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### **IONIZATION TECHNIQUES**

### • ION SOURCE

**Spec** 





- Electron impact (EI)
- Chemical Ionization (CI)
- Atmospheric Pressure Ionization (API)
  - •Electrospray Ionization (ESI)
  - •Atmospheric Pressure Chemical Ionization (APCI)
  - •Atmospheric Pressure Photo-Ionization (APPI)
- Matrix Assisted Laser Desorption/Ionization (MALDI)



#### Three Fundamental Processes:

- 1. Production of charged droplets.
- 2. Droplet size reduction, and fission.
- 3. Gas phase ion formation.





## Ion Evaporation Theory





#### ESI - Ion Max Source







#### **Electrospray Ionization**

#### Atmospheric Pressure Chemical ionization

## Chemistry Considerations ESI or APCI

#### ESI:

- lons formed by solution chemistry
- Good for thermally labile analytes
- Good for polar analytes
- Good for large molecules (Proteins / Peptides)

#### APCI:

- lons formed by gas phase chemistry
- Good for volatile / thermally stable
- Good for non-polar analytes
- Good for small molecules (Steroids)



#### Ion Max Source Design - APCI Probe





- It depends on the exact application.
- Increasing polarity and molecular weight and thermal instability favors electrospray.
  - Most drugs of abuse are highly polar and are easily analyzed using electrospray.
  - High molecular weight proteins also require electrospray
- Lower polarity and molecular weight favors APCI or APPI.
  - Lower background, but compounds must be more thermally stable.





## Typical Mass Accuracy and Resolution

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Type of MS	Mass accuracy	Resolution	Utility for
Quadrupole	0.1 amu	6,000	Identify
Traps	0.1 amu	8,000	Identify
TOF	0.0001 amu	<20,000 TOF	Empirical formula/
		60,000 Q-TOF	composition
Sector	0.0001 amu	10,000	Empirical formula/
			composition
Orbitrap	0.0001 amu	1,000,000	Empirical formula/
			composition



## ---- MASS ANALYSER

#### QUADRUPLE

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![](_page_28_Figure_1.jpeg)

## **Sci** TSQ Triple Quadrupole (available on YouTube)

![](_page_29_Picture_1.jpeg)

http://www.youtube.com/watch?v=LFB14D8pkoc

![](_page_30_Picture_0.jpeg)

Scan Mode	Q1	Q2	Q3	Purpose
Full Scan	Scanning	Pass All	Pass All	MW Info.
SIM	Fixed m/z	Pass All	Pass All	Quantitation
Product	Fixed m/z	Pass All (+ CE)	Scanning	Structural Info.
SRM	Fixed m/z	Pass All (+ CE)	Fixed m/z	Targeted Quantitation
Neutral Loss	Scanning	Pass All (+ CE)	Scanning	Analyte Screening
Precursor	Scanning	Pass All (+ CE)	Fixed m/z	Analyte Screening

![](_page_31_Picture_0.jpeg)

### Full Scan Mode

![](_page_31_Figure_2.jpeg)

![](_page_32_Picture_0.jpeg)

## Full Scan (Q1 or Q3)

#### Full Scan Mode

#### Purpose: Survey scan of a chromatographic peak

![](_page_32_Figure_4.jpeg)

Q1 RF Only RF Only Q3 Scanning

Full Scan Q3:

![](_page_33_Picture_0.jpeg)

## Selected Ion Monitoring = SIM

#### SIM Mode

#### Purpose: Quantitation on a specific m/z range of ions

![](_page_33_Figure_4.jpeg)

![](_page_34_Picture_0.jpeg)

## Selected Ion Monitoring = SIM

SIM is in essence a full scan acquisition on a relatively narrow mass window (defined as center mass / scan width)

![](_page_34_Figure_3.jpeg)

- -- Advantages
  - □ Targeted analyte monitoring
  - ◻ High duty cycle

- Disadvantages
  - $\propto$  Can suffer from interferences
  - lpha Not as sensitive or selective as SRM

![](_page_35_Figure_0.jpeg)

![](_page_35_Figure_1.jpeg)

## Selected Reaction Monitoring (SRM)

![](_page_36_Figure_1.jpeg)

- -- Advantages
  - □ Targeted analyte monitoring
  - ◻ High duty cycle
  - Simultaneous" monitoring of multiple transitions

- Disadvantages

![](_page_36_Picture_8.jpeg)

![](_page_37_Figure_0.jpeg)

## The Need for True MS/MS

![](_page_37_Figure_2.jpeg)

![](_page_38_Picture_0.jpeg)

#### SRM Selectivity in Complex Matrices

RT: 2.28 - 5.89 SM: 15G

![](_page_38_Figure_3.jpeg)

![](_page_39_Figure_0.jpeg)

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![](_page_39_Figure_1.jpeg)

### HIGH RESOLUTION MASS ANALYSER

![](_page_40_Picture_1.jpeg)

Spec

![](_page_40_Picture_2.jpeg)

![](_page_41_Picture_0.jpeg)

#### Nominal Mass

The mass of an ion with a given empirical formula calculated using the integer mass numbers of the most abundant isotope of each element

Ex: M=249  $C_{20}H_{9}^{+}$  <u>or</u>  $C_{19}H_{7}N^{+}$  <u>or</u>  $C_{13}H_{19}N_{3}O_{2}^{+}$ 

#### • Exact Mass

Ех

The mass of an ion with a given empirical formula calculated using the exact mass of the most abundant isotope of each element

: M=249	$C_{20}H_{9}+$	249.0070
	$C_{19}H_7N+$	249.0580
	$C_{13}H_{19}N_3O_2+$	249.1479

![](_page_42_Picture_0.jpeg)

Sci Spec

![](_page_42_Picture_1.jpeg)

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## Mass Resolution: What is it?

- Ability of a mass spectrometer to distinguish between ions of nearly equal m/z ratios (isobars).
  - m measured mass
  - ▲m peak width measured at 50% peak intensity (Full Width Half Maximum)
    - or the mass difference
      between two adjacent peaks
      of equal intensity, in this case
      pw @ 10% valley definition is
      used.

![](_page_43_Picture_6.jpeg)

![](_page_43_Figure_7.jpeg)

![](_page_44_Picture_0.jpeg)

### **Resolution & Peak Width**

![](_page_44_Figure_2.jpeg)

![](_page_45_Picture_0.jpeg)

- At minimum the resolution of the mass analyzer should be sufficient to separate two ions differing by one mass unit anywhere in the mass range scanned (unit mass resolution).
- Typical values of resolution for <u>low resolution mass analyzers</u> (e.g. quadrupoles and ion traps) are below 5000.
- <u>High resolution instruments</u> have a resolution exceeding 15000.

## Sci Commercial High Resolution MS Technology Race

![](_page_46_Figure_1.jpeg)

Mass resolution (FWHM)

![](_page_47_Picture_0.jpeg)

Anal. Chem. 2000, 72, 1156-1162

#### Electrostatic Axially Harmonic Orbital Trapping: A High-Performance Technique of Mass Analysis

#### Alexander Makarov\*

HD Technologies Ltd., Atlas House, Simonsway, Manchester, M22 5PP, U.K.

![](_page_47_Picture_5.jpeg)

Major accurate-mass analyzers for Life Science Sci Spec FT ICR TOF MS **Orbitrap MS**  Excitation by injection Motion in preions are trapped Motion in electro- Detection by dominantly magnetic Image current detection

- field
- Low energy injection

 $T \propto m/z$ 

- MS<sup>n</sup> possibilities
- Broad-band excitation

- Fourier transform and
- data processing
- Significant kinetic energy during detection
- Very long mean free path (many km)

- static fields (m/zindependent well)
- High energy injection
- High energy spread upon fragmentation
- secondary electron multiplier
- Very high kinetic energy for detection (many kV)
- Significant mean free path (tens of m)

 $TOF \propto \sqrt{m/z}$ 

 $T \propto \sqrt{m/z}$ 

# Orbitrap Mass Analyzer: Principle of Operation

![](_page_49_Figure_1.jpeg)

Makarov A. Anal. Chem. 2000, 72, 1156-1162.

#### Many Ions Generate a Complex "Transient"

![](_page_50_Figure_1.jpeg)

![](_page_51_Picture_0.jpeg)

#### UHPLC with Q Exactive Mass Spectrometer

![](_page_51_Picture_2.jpeg)

![](_page_51_Picture_3.jpeg)

http://planetorbitrap.com/

#### **ThermoFisher** SCIENTIFIC

## Applications of Triple Quadrupole LC-MS/MS

![](_page_53_Picture_0.jpeg)

#### thermoscientific

![](_page_53_Picture_3.jpeg)

Robustness, reproducibility, reliability with best-in-class sensitivity: Increased confidence in targeted quantitation of pesticides in food matrices

![](_page_54_Picture_0.jpeg)

## Quantitation of more than 250 Pesticides below MRLs in Leek

![](_page_54_Picture_3.jpeg)

LC: Vanquish Flex Binary System

Column: Accucore aQ (2.1 × 100 mm, 2.6 µm)

Column Temperature: 25°C

Injection Volume: 1 µL

Mobile Phase: A) 98% water with 2% methanol; B) 98% methanol with 2% water— Both containing 0.1% formic acid and 5 mM ammonium formate

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Flow Rate: 300 µL/min

Run Time: 15 min

![](_page_55_Picture_0.jpeg)

![](_page_55_Picture_2.jpeg)

- Maximize UHPLC separation with 1034 bar (15,000 psi) pump pressure limit
- Viper-based, tool-free fluidic connections
- Biocompatible, Iron-free flow path
- Sample pre-compression for better injection reproducibility and longer column lifetimes
- Standard Autosamper capacity: 4 racks (216 vials)
- New column thermostatting technology
- Removable doors for easy access

![](_page_56_Picture_0.jpeg)

#### **TSQ Quantis Parameter**

![](_page_56_Picture_2.jpeg)

lonization mode	Heated Electrospray (HESI)
Scan type	timed-SRM
Polarity	Positive/Negative switching
Spray Voltage for Positive mode	3700 V
Spray Voltage for Negative mode	2500 V
Sheath gas pressure	30 arbitrary units (Arb)
Aux gas pressure	6 Arb
Sweep gas pressure	1 Arb
lon transfer tube temperature	325 °C
Vaporizer temperature	350 °C
CID gas pressure	2 mTorr
Cycle time	0.5 s
Q1 resolution (FWHM)	0.7
Q3 resolution (FWHM)	0.7

![](_page_57_Picture_0.jpeg)

![](_page_57_Figure_2.jpeg)

![](_page_58_Picture_0.jpeg)

#### Chromatogram of more than 250 pesticides

![](_page_58_Figure_2.jpeg)

LC-MS/MS chromatogram of more than 250 pesticides in leek extract at 100  $\mu$ g/kg.

![](_page_59_Picture_0.jpeg)

#### Azoxystrobin elutes at 8.69 min

![](_page_59_Figure_2.jpeg)

![](_page_60_Figure_0.jpeg)

#### TSQ Quantis MS *VS* TSQ Endura MS

![](_page_60_Figure_2.jpeg)

Differences in performance are shown in peak area and peak height.

## Sci Reliable Performance when Starting from Standby mode

![](_page_61_Figure_1.jpeg)

- Red lines represent -20% of Atrazine response at 10  $\mu$ g/kg.
- Yellow lines show the exact moment the system was placed in standby mode for 12 h (no maintenance was performed).
- The data shows that the response was within the expected —20% range for at least 400 injections of 10 ppb QC in leek.

![](_page_62_Picture_0.jpeg)

- rapid and robust quantitation of more than 250 pesticides in leek at or below their respective MRLs.
- selectivity and sensitivity enabled analysis of only 1 μL sample
- without need for dispersive SPE sample cleanup or sample dilution.

![](_page_62_Picture_5.jpeg)

![](_page_63_Picture_0.jpeg)

#### Thank You for Your Attention

## Get Connected www.scispec.co.th/

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![](_page_63_Figure_4.jpeg)

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![](_page_63_Picture_6.jpeg)

f

https://www.facebook.com/scispec/

![](_page_63_Picture_9.jpeg)

![](_page_63_Picture_10.jpeg)